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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO	
09/807,658	08/27/2001	Kazuya Nakada	M 6820 PCT/US	2363	
7:	590 06/13/2003				
Henkel Corporation			EXAMINER		
2500 Renaissar Gulph Mills, P.	nce Boulevard Suite 200 A 19406		OLTMANS, ANDREW L		
			ART UNIT	PAPER NUMBER	
			1742		
		DATE MAIL ED: 06/12/2002			

Please find below and/or attached an Office communication concerning this application or proceeding.

			R				
V-	Application No.	Applicant(s)					
	09/807,658	NAKADA, KAZUY	'A				
Office Action Summary	Examiner	Art Unit					
	Andrew L Oltmans	1742					
The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply							
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a reply If NO period for reply is specified above, the maximum statutory period w Failure to reply within the set or extended period for reply will, by statute, - Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b). Status	B6(a). In no event, however, many within the statutory minimum or ill apply and will expire SIX (6) cause the application to become	ay a reply be timely filed of thirty (30) days will be considered timel MONTHS from the mailing date of this cone ABANDONED (35 U.S.C. § 133).					
1) Responsive to communication(s) filed on 03 A	pril 2003 .						
2a)⊠ This action is FINAL . 2b)□ Thi	s action is non-final.		• .				
3) Since this application is in condition for allowa closed in accordance with the practice under <i>B</i> Disposition of Claims			ne merits is				
4)⊠ Claim(s) 1 and 10-36 is/are pending in the app	lication.						
4a) Of the above claim(s) is/are withdrawn from consideration.							
5) Claim(s) is/are allowed.							
6)⊠ Claim(s) <u>1 and 10-36</u> is/are rejected.							
7) Claim(s) is/are objected to.							
8) Claim(s)are subject to restriction and/or	election requirement						
Application Papers							
9)☐ The specification is objected to by the Examiner							
10)☐ The drawing(s) filed on is/are: a)☐ accepted or b)☐ objected to by the Examiner.							
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).							
11) ☐ The proposed drawing correction filed on is: a) ☐ approved b) ☐ disapproved by the Examiner.							
If approved, corrected drawings are required in reply to this Office action.							
12) The oath or declaration is objected to by the Exa	aminer.						
Priority under 35 U.S.C. §§ 119 and 120	•						
13) Acknowledgment is made of a claim for foreign	priority under 35 U.S	.C. § 119(a)-(d) or (f).					
a)⊠ All b)□ Some * c)□ None of:	•						
1. Certified copies of the priority documents							
2. Certified copies of the priority documents	have been received	in Application No					
3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received.							
14) Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).							
a) ☐ The translation of the foreign language prov 15)☐ Acknowledgment is made of a claim for domestic	* *	•					
Attachment(s)	· •						
Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449) Paper No(s)	5) Notice	riew Summary (PTO-413) Raper Nor e of Informal Patent Application (PT :					

Art Unit: 1742

DETAILED ACTION

Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Sako et al. 4,954,372 in view of Japanese Patent JP 06-116527 A

2. Claims 1, 10-36 are rejected under 35 U.S.C. 103(a) as being unpatentable over Sako et al. 4,954,372 (Sako; cited on IDS filed December 7, 2001) in view of Japanese Patent JP 06-116527 A (JP '527; cited on IDS filed December 7, 2001).

Sako teaches a coated hydrophilic metallic material, a method of forming the coated material and the composition used in the process of forming the coated material wherein the coated material is useful as a heat exchanger and the method of coating includes a step of degreasing, pre-treating with a conversion coating, coating with a hydrophilic composition, and drying, as recited in claims 14-16, 28-29, 31-32 (col 1, lines 7-14; col 4, lines 17, col 8, lines 63-66). Sako teaches that the coating composition includes a first polymer (corresponding to component A instantly claimed), a second polymer (corresponding to component B instantly claimed) and a crosslinking agent, as recited in claim 1, as recited in claims 1, 10-11, 20-26 and 35-36 (col 4):

Art Unit: 1742

(a) a first water-soluble polymer having a chemical structure formally obtainable by polymerizing a mixture of molecules all containing addition-polymerizable carbon-carbon double bonds, wherein at least 40 number percent of the molecules in the mixture have a structure according to the general formula (I)

$$\begin{array}{c|c}
R_1 & O \\
\downarrow & \parallel \\
H_2C = C - C - N
\end{array}$$

$$\begin{array}{c|c}
R_2 \\
R_3
\end{array}$$
(I)

wherein R₁ represents H or CH₃; R₂ and R₃ each independently represents H, alkyl having 1 to 4 carbon atoms, benzyl, or hydroxyalkyl having 2 or 3 carbon atoms;

- (b) a second water-soluble polymer having a chemical structure formally obtainable by polymerizing a mixture of molecules all containing addition-polymerizable carbon-carbon double bonds, wherein at least 60 number percent of the molecules in the mixture also contain a moiety selected from the group consisting of carboxyl groups, carboxylate salts, sulfonic acid group, sulfonate salts, phosphonic acid group, phosphonate salts, amino groups, salts of amino groups, and quaternary ammonium groups; and
 - (c) a water-soluble crosslinking agent which is cosoluble with both (a) and (b).

[see also col 4, line 55 to col 6, line 37]

Sako teaches that crosslinking agent may include a metal ion, including vanadium, titanium and zirconium (col 7, lines 4-32). Sako also teaches the addition of the ingredient recited in claim 12 (col 8, lines 46-51). Sako teaches compositional ranges that appear to overlap those instantly claimed in claims 13, 17-19, 27 and 34 (col 7, lines 50-54; col 8, lines 6-26 and Sako: claim 10).

Application/Control Number: 09/807,658

Art Unit: 1742

Sako teaches a thickness for the hydrophilic coating that encompasses the range instantly claimed in claims 30 and 33 (col 8, lines 59-60).

Although Sako teaches the inclusion of a crosslinking agent selected from a list of metals including vanadium, titanium and zirconium, Sako fails to meet all the limitations of the instant claims in that Sako does not explicitly teach the combination of vanadium with titanium, zirconium or silicon. Sako also fails to teach the exact same ranges of composition.

JP '527 teaches crosslinking agents for hydrophilic polymer compositions, wherein the crosslinking agents include vanadium (see abstract), with compounds including silicon (see abstract), titanium and zirconium (see paragraph [0013] of English language translation). JP '527 teaches that the combination of vanadium with titanium (fluoride), zirconium (fluoride) or silicon (silica or fluoride) work together to provide the hydrophilic coating with excellent endurance and corrosion resistance (see paragraph [0013] of English language translation).

One of ordinary skill in the art at the time that the invention was made would have found the selection of vanadium in combination with titanium, zirconium or silicon as the crosslinking agents of Sako obvious because one of ordinary skill in the art would have been motivated to provide Sako with the desirable properties that result when vanadium is used together with titanium (fluoride), zirconium (fluoride) or silicon (silica or fluoride), including excellent endurance and corrosion resistance, as taught in JP '527.

With respect to the compositional ranges, one of ordinary skill in the art at the time the invention was made would have considered the invention to have been obvious because the coating compositions taught by the reference overlap that of the instant claims, <u>In re Malagari</u>, 182 USPQ 549, and MPEP 2144.05.

Art Unit: 1742

Response to Arguments

3. Applicant's arguments filed April 4, 2003 have been fully considered but they are not persuasive. Claims 1 and 10-36 remain pending in this application. The rejections made in the previous Office Action have been maintained and the newly presented claims (i.e. 35-36) have been rejected, as appropriate.

- 4. With respect to the limitations that removes Si from group (D), the claims remain obvious because the JP '527 reference teaches Zr and Ti. JP '527 teaches that the combination of vanadium with titanium (fluoride), zirconium (fluoride) or silicon (silica or fluoride) work together to provide the hydrophilic coating with excellent endurance and corrosion resistance (see paragraph [0013] of English language translation). It is further noted, contrary to applicant's arguments, that the particular form of the Ti and Zr does not distinguish over the claimed inclusion of Zr and Ti. The claims merely recite the inclusion of Ti or Zr.
- 5. With respect to applicant's argument that there is no motivation to combine references, as recited in the previous rejection, one of ordinary skill in the art would have been *motivated* to provide Sako with the desirable properties that result when vanadium is used together with titanium (fluoride), zirconium (fluoride) or silicon (silica or fluoride), including excellent endurance and corrosion resistance, as taught in JP '527, by adding the complexing agent taught therein. Therefore, applicant's assertion that there is no motivation to combine is not persuasive. Likewise, the argument that impermissible hindsight was used is not persuasive because the motivation comes from the references, not from applicant's disclosure.

Art Unit: 1742

6. With respect to claim 35, the polymer (A) recited therein do not distinguish over the teachings of the Sako (see rejection, above).

- 7. In response to applicant's arguments against the Sako individually, including the discussion of the number of variations of the complexing agent, one cannot show nonobviousness by attacking references individually where the rejections are based on combinations of references. See *In re Keller*, 642 F.2d 413, 208 USPQ 871 (CCPA 1981); *In re Merck & Co.*, 800 F.2d 1091, 231 USPQ 375 (Fed. Cir. 1986). In this case, the complexing agent is suggested by the secondary reference JP '527, wherein motivation is provided to combine the disclosures (see paragraph 5, above).
- 8. With respect to applicant's argument that the composition of Sako and JP '527 are "completely different", the applicant has not specifically laid out reasoning for the differences, other than noting that JP '527 contains silicon. The examiner notes that both compositions are hydrophilic polymer containing compositions utilized for coating metallic surfaces. In view of the similar composition and use, one of ordinary skill in the art would find a reasonable expectation of success in the combination. Therefore, for the reasons set forth in the previous Office Action, the claimed invention is obvious.
- 9. With respect to the limitation "free of chromium", the claims do not unobvious over the teachings of Sako and JP '527 because Sako and JP '527 do not require Cr. JP '527 does not suggest the addition of chromium at all. Although Sako teaches chromium, as pointed out by applicant with respect to vanadium, titanium and zirconium, chromium is one of a list of elements that can be used (see Table 1). Although chromium is taught in the embodiments, the reference is not limited to the preferred embodiments, but rather is read as a whole, MPEP 2123:

Application/Control Number: 09/807,658

Art Unit: 1742

"The use of patents as references is not limited to what the patentees describe as their own inventions or to the problems with which they are concerned. They are part of the literature of the art, relevant for all they contain." In re Heck, 699 F.2d 1331, 1332-33, 216 USPQ 1038, 1039 (Fed. Cir. 1983) (quoting In re Lemelson, 397 F.2d 1006, 1009,158 USPQ 275, 277 (CCPA 1968)).

A reference may be relied upon for all that it would have reasonably suggested to one having ordinary skill the art, including nonpreferred embodiments. Merck & Co. v. Biocraft Laboratories, 874 F.2d 804, 10 USPQ2d 1843 (Fed. Cir.), cert. denied, 493 U.S. 975 (1989).

If the reference is read as a whole, the composition free of chromium is taught, therefore making the instant invention obvious.

Conclusion

10. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

11. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Andrew L Oltmans whose telephone number is 703-308-2594. The examiner can normally be reached 7:00-3:30 Monday-Friday.

Art Unit: 1742

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Roy King can be reached on 703-308-1146. The fax phone numbers for the organization where this application or proceeding is assigned are 703-872-9310 for regular communications and 703-873-9311 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703-308-0661.

ALO

June 10, 2003

John P. Sheehan
Primary Examiner

Art Unit 1742